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Spectral Parameter Calculation of Ho^{3+} , Tm^{3+} Co-doped KYb(WO_4)₂ Laser Crystal

BI Lin^{1*}, DI Xiao-qiang¹, ZHAO Jian-ping¹, LI Chun², LIU Jing-he²

 School of Computer Science and Technology, Changchun University of Science and Technology, Changchun 130022, China;
 School of Materials Science and Engineering, Changchun University of Science and Technology, Changchun 130022, China) * Corresponding Author, E-mail; bilin7080@163.com

Abstract: $\text{KHo}_{0.04} \text{Tm}_{0.06} \text{Yb}_{0.9} (WO_4)_2$ laser crystal was grown by the top seeded solution growth (TSSG) method. The absorption and fluorescent spectra of the crystal were measured, and the spectral parameters were calculated, too. The experimental results show that there is a broader absorption band at the range of 890 – 1 000 nm with a full width at half maximum (FWHM) of 90 nm. The absorption cross section of main emission peak at 1 000 nm is $16.92 \times 10^{-20} \text{ cm}^2$. Meanwhile, Tm^{3+} ions in the crystal have a wider absorption band at 1 690 – 1 812 nm with a FWHM of 118 nm. Therefore, the energy transfer of Yb \rightarrow Ho, Yb \rightarrow Tm, Tm \rightarrow Ho can be easily achieved. The spectral strength parameters of the crystal were calculated based on the Judd-Ofelt theory. According to the energy level diagram of Tm³⁺, Ho³⁺ and Yb³⁺ ions, three kinds of energy transfer modes at 1 750 – 2 200 nm emission were discussed. The stimulated emission cross section of main emission peak at 2 030 nm is $3.47 \times 10^{-20} \text{ cm}^2$, which indicates that the crystal can be used as an excellent laser gain medium at the range of 2 µm.

Key words: $KHo_{0.04}Tm_{0.06}Yb_{0.9}(WO_4)_2$ laser crystal; absorption spectrum; fluorescent spectrum; spectral parameter calculation

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钬铥双掺钨酸镱钾激光晶体光谱参数计算

毕 琳¹*,底晓强¹,赵建平¹,李 春²,刘景和² (1. 长春理工大学 计算机科学技术学院,吉林长春 130022; 2. 长春理工大学 材料科学与工程学院,吉林长春 130022)

摘要:采用顶部籽晶提拉法(TSSG)生长了钬铥双掺钨酸镱钾(KHo_{0.04} Tm_{0.06} Yb_{0.9}(WO₄)₂)激光晶体。测试 了该晶体的吸收及荧光光谱,计算了其光谱参数。实验结果表明:该晶体在 890 ~1 000 nm 范围吸收带较宽, 半峰宽为 90 nm,计算了主峰1 000 nm 处吸收截面为 16.92 × 10⁻²⁰ cm²;Tm³⁺在 1 690 ~1 812 nm 范围存在较 宽的吸收带,半峰宽为 118 nm,易于实现 Yb→Ho、Yb→Tm、Tm→Ho 的能量传递。根据 Judd-Ofelt 理论,计算 了该晶体的光谱强度参数。根据 Tm³⁺、Ho³⁺、Yb³⁺离子能级图,讨论了产生 1 750 ~2 200 nm 荧光发射的 3 种 能量传递方式。最后计算了主峰 2 030 nm 处受激发射截面为 3.47 × 10⁻²⁰ cm²,表明该晶体可作为 2 μ m 波段 优异的激光增益介质。

关 键 词: KHo_{0.04}Tm_{0.06}Yb_{0.9}(WO₄)2激光晶体;吸收光谱;荧光光谱;光谱参数计算

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1 Introduction

Ho³⁺, Tm³⁺ ions can be commonly used as the activation ions at the range of $2-3 \mu m$ wavelength, which has the advantage of large stimulated-emission cross-section and long fluorescence lifetime. In the military, 2 μm laser has a strong ability to penetrate the smoke of the battlefield with good confidentiality. These two types of activation ions are suitable for military laser ranging, laser radar and electro-optical jamming, $etc^{[14]}$. In the medical field, 2 μm laser can perform precise excision of biological tissues without damaging surrounding tissues, while reducing the anesthetic dosage^[5]. In addition, it can be used to detect harmful gases such as formaldehyde with the magnitude of 10^{-9} , as well as carbon monoxide in the environment^[6].

 ${\rm Tm}^{3+}$ ions can be often used as the sensitizing agent of Ho³⁺ ions since Tm³⁺ ions have stronger absorption efficient than Ho³⁺ ions at the range of 785 – 810 nm. It can achieve the energy transfer of Tm \rightarrow Ho and then improve the absorption efficiency to pumping light^[7-11]. Yb³⁺ ions have been widely used as a sensitizing agent with the LD pumping popular. The absorption peak of Yb³⁺ ions located near 980 nm and can be effectively coupled with InGaAs LD. Because the energy level structure of Yb³⁺ ion is very simple without excitation state reabsorption process, Yb³⁺ can be served as the sensitizing agent of Ho³⁺, Tm³⁺ ions.

KYb(WO_4)₂ is one of the self-activation laser crystals with a double tungstates structure. In the host of KYb(WO_4)₂, Yb³⁺ ions have been served as not only an integral part of the host, but also the active ions. The absorption wavelength of KYb(WO_4)₂ is 930 – 980 nm, which can be effectively coupled with the InGaAs LD pumping^[12-14]. The laser output wavelength of KYb(WO_4)₂ is in the range of 980 – 1 080 nm. Laser emission with multi-wavelength and tenability in this crystal can be achieved by doping Er^{3+} , Tm^{3+} , Ho^{3+} , *etc. Re*: KYb(WO_4)₂ has several merits including high doping concentration, low fluorescence quenching, high conversion efficiency, low threshold value and well stability, *etc.* It can achieve all-solid-state, miniaturization and integration for lasers and widely used in the fields of human-eye safe, laser communication, medical treatment, remote sensing, $etc^{[15-18]}$. In this paper, the growth, absorption and fluorescence spectra of KHo_{0.04}Tm_{0.06}-Yb_{0.9} (WO₄)₂ laser crystal were investigated, and the spectral parameters of this crystal were calculated as well.

2 Experiments

2.1 Crystal Growth

The chemicals included K_2CO_3 (top-grade pure), $Yb_2O_3(99.99\%)$, $Ho_2O_3(99.99\%)$, $Tm_2O_3(99.99\%)$, and WO_3 (99.999\%). The $K_2W_2O_7$ was served as solvent. They were mixed and reacted based on the equations below:

 $0.04 \text{Ho}_2\text{O}_3 + 0.06 \text{Tm}_2\text{O}_3 + 0.9 \text{Yb}_2\text{O}_3 + \text{K}_2\text{CO}_3 + 4\text{WO}_3 \rightarrow 2\text{KHo}_{0.04} \text{Tm}_{0.06} \text{Yb}_{0.9} (\text{WO}_4)_2 + \text{CO}_2,$ (1)

$$K_2CO_3 + 2WO_3 \rightarrow K_2W_2O_7 + CO_2, \quad (2)$$

where the mole ratio of $KHo_{0.04}Tm_{0.06}Yb_{0.9}(WO_4)_2$ to $K_2 W_2 O_7$ was 1:4. The crystal was grown by the Top Seeded Solution Growth (TSSG) method. The experimental equipment included a MCGE-furnace heated by resistance wire, a platinum crucible with a size of $\Phi 60 \text{ mm} \times 50 \text{ mm}$, a Pt-Rh thermocouple, and an AI-808P thermal controller. The mixed materials was put in a platinum crucible and fully melted at a temperature of 80 °C above super-saturation for 24 h. The oriented crystallon was dipped at about 910 ℃, and grown gradually with a rotation rate of 15 r/min, a pulling rate of 2 mm/day and a cooling rate of 0.05/h. After 2 weeks, the crystal was removed from the crucible and cooled down to the room temperature at a rate of 20 °C/h. Finally, the KHo_{0.04}Tm_{0.06}Yb_{0.9} (WO₄)₂ laser crystal with a dimension of 26 mm × 11 mm × 9 mm was obtained successfully. Then the crystal sample with a dimension of 3 mm × 3 mm × 1 mm was cut perpendicularly to b axis for spectral measurement.

2.2 Measurement of Sample

The absorption spectrum of the crystal sample was measured using an ultraviolet (UV) spectrophotometer(Model UV360, Shimadzu Company, Japan) at room temperature. The fluorescence spectrum of the crystal sample was measured with an X-ray fluorescence spectrometer (FLUOROLOG-3, HORIBA Company, Japan) at room temperature.

3 Results and Discussion

3.1 Absorption Spectrum

The absorption spectrum of the crystal sample is shown in Fig. 1. There are several absorption peaks at 378, 421, 450, 475, 538, 625, 645, 691, 805, 890, 916, 980, 1 000, 1 190, 1 223, 1 690. 1712, 1749, 1804, 1812, 1924 nm. It can be seen that the absorption peaks at 378, 475, 805 and 1 690 – 1 812 nm are accounting for ${}^{3}H_{6} \rightarrow {}^{1}D_{2}$ (Tm³⁺), ${}^{3}\text{H}_{6} \rightarrow {}^{1}\text{G}_{4}(\text{Tm}^{3+}), {}^{3}\text{H}_{6} \rightarrow {}^{3}\text{H}_{4}(\text{Tm}^{3+}) \text{ and } {}^{3}\text{H}_{6} \rightarrow {}^{3}\text{F}_{4}$ (Tm³⁺) transitions, respectively. Meanwhile, the absorption peaks located at 421, 450, 538, 625 -691, 890, 916, 1 190 (1 223), 1 924 nm are due to ${}^{5}I_{8} \rightarrow {}^{5}G_{5}$ (Ho³⁺), ${}^{5}I_{8} \rightarrow {}^{5}G_{6}$ (Ho³⁺), ${}^{5}I_{8} \rightarrow {}^{5}F_{4}$ $(\mathrm{Ho}^{3+}), {}^{5}\mathrm{I}_{8} \rightarrow {}^{5}\mathrm{F}_{5}(\mathrm{Ho}^{3+}), {}^{5}\mathrm{I}_{8} \rightarrow {}^{5}\mathrm{I}_{4}(\mathrm{Ho}^{3+}), {}^{5}\mathrm{I}_{8} \rightarrow {}^{5}\mathrm{I}_{5}$ $(\mathrm{Ho}^{3+}), {}^{5}\mathrm{I}_{8} \rightarrow {}^{5}\mathrm{I}_{6}(\mathrm{Ho}^{3+}), {}^{5}\mathrm{I}_{8} \rightarrow {}^{5}\mathrm{I}_{7}(\mathrm{Ho}^{3+})$ transitions, respectively. The absorption peaks at 981 and 1 000 nm are attributed to ${}^{2}F_{7/2} \rightarrow {}^{2}F_{5/2}$ (Yb³⁺) transition. The full width at half maximum (FWHM) at the range of 890 - 1 000 nm amounts to 90 nm. which is more propitious for InGaAs LD laser pumping. In this absorption spectrum, the most interesting is a broader absorption band ($\mathrm{Tm}^{^{3\,+}}$) at 1 690 – 1 812 nm with a FWHM of 118 nm. The effective energy transfer of Yb→Ho, Yb→Tm, Tm→Ho can be easily achieved to produce 2 µm around laser due to these two broader absorption bands at 890 -



Fig. 1 Absorption spectrum of $\rm KHo_{0.04}\,Tm_{0.06}\,Yb_{0.9}\,(\,WO_{4}\,)_{\,2}$ crystal

 $1~000~nm(\,{\rm Yb}^{3\,*}\,)$ and $1~690-1~812~nm(\,{\rm Tm}^{3\,*}\,)$ in the crystal.

The absorption cross section σ_{abs} of the crystal can be defined as formula (3)^[19]:

$$\sigma_{\rm abs} = \frac{\int k(\lambda) \, \mathrm{d}\lambda}{N} = \frac{D(\lambda)}{\lg \cdot NL}, \qquad (3)$$

where *D* is the optical density, which can be calculated as $D = \lg(I_0/I)$, based on absorption spectrum measurements; I_0 and *I* represent the intensities of incident light and emergent light, respectively; *N* represents the concentration of Ho³⁺, Tm³⁺ or Yb³⁺ ions, and *L* represents the thickness of the sample (L = 1 mm). The absorption cross section of the crystal at the strongest peak of 1 000 nm was calculated as 16. 92 × 10⁻²⁰ cm². Tab. 1 lists the main absorption peaks and absorption parameters of the crystal.

Meanwhile, the spectral intensity parameters in the crystal can be obtained based on the Judd-Ofelt theory^[20-21] as follows: $\Omega_2 = 10.24 \times 10^{-20} \text{ cm}^2$, $\Omega_4 = 2.37 \times 10^{-20} \text{ cm}^2$, $\Omega_6 = 1.43 \times 10^{-20} \text{ cm}^2$. **Tab. 1** Main absorption peaks and absorption parame-

ters of the $KHo_{0.04}Tm_{0.06}Yb_{0.9}(WO_4)_2$ crystal

Transition	λ/nm	N/cm^{-3}	$\sigma_{ m abs}/ m cm^2$
$\overline{{}^{3}\mathrm{H}_{6}} \rightarrow {}^{1}\mathrm{D}_{2}(\mathrm{Tm}^{3+})$	378	2.94×10^{20}	1.17×10^{-20}
${}^{5}\mathrm{I}_{8}{\rightarrow}{}^{5}\mathrm{G}_{5}(\mathrm{Ho}^{3+})$	421	1.96×10^{20}	0.92×10^{-20}
${}^{5}\mathrm{I}_{8}{\rightarrow}{}^{5}\mathrm{G}_{6}(\mathrm{Ho}^{3+})$	450	1.96×10^{20}	3.38×10^{-20}
${}^{3}\mathrm{H}_{6} \rightarrow {}^{1}\mathrm{G}_{4}(\mathrm{Tm}^{3+})$	475	2.94×10^{20}	2.12 × 10 $^{-20}$
$^{5}\mathrm{I}_{8}{\rightarrow}^{5}\mathrm{F}_{4}(\mathrm{Ho}^{3+})$	538	1.96×10^{20}	3.89×10^{-20}
${}^{5}\mathrm{I}_{8}{\rightarrow}{}^{5}\mathrm{F}_{5}(\mathrm{Ho}^{3+})$	625	1.96×10^{20}	1.93×10^{-20}
	645	1.96×10^{20}	1.95×10^{-20}
	691	1.96×10^{20}	4.47 × 10 $^{-20}$
$^{3}\mathrm{H}_{6}{\rightarrow}^{3}\mathrm{H}_{4}(\mathrm{Tm}^{3+})$	805	2.94×10^{20}	3.28×10^{-20}
$^5\mathrm{I}_8{\longrightarrow}^5\mathrm{I}_4(\mathrm{Ho}^{3+})$	890	1.96×10^{20}	5.32×10^{-20}
$^5\mathrm{I}_8{\rightarrow}^5\mathrm{I}_5(\mathrm{Ho}^{3+})$	916	1.96×10^{20}	7.89 × 10 $^{-20}$
$^2\mathrm{F}_{5/2}{\longrightarrow}^2\mathrm{F}_{5/2}(\mathrm{Yb}^{3+})$	980	4.4×10^{21}	15.27×10^{-20}
	1 000	4.4×10^{21}	16.92×10^{-20}
$^5\mathrm{I}_8{\longrightarrow}^5\mathrm{I}_6(\mathrm{Ho}^{3+})$	1 190	1.96×10^{20}	5.31 × 10 $^{-20}$
	1 223	1.96×10^{20}	5.46 × 10 $^{-20}$
${}^{3}\text{H}_{6} \rightarrow {}^{3}\text{F}_{4} (\text{Tm}^{3+})$	1 690 - 1 812	2.94×10^{20}	10.35 × 10 $^{-20}$
$^{5}\mathrm{I}_{8}{\rightarrow}^{5}\mathrm{I}_{7}(\mathrm{Ho}^{3+})$	1 924	1.96×10^{20}	3.02×10^{-20}

3.2 Fluorescence Spectrum

The fluorescence spectrum of the crystal sample

under 980 nm pumping is shown in Fig. 2. There are four obvious emission peaks at 950 – 1 050 nm, 1 100 – 1 250 nm, 1 400 – 1 470 nm, 1 750 – 2 200 nm, which are attributed to ${}^{2}F_{5/2} \rightarrow {}^{2}F_{7/2}$ (Yb³⁺), ${}^{3}H_{5} \rightarrow {}^{3}H_{6}$ (Tm³⁺) & ${}^{5}I_{6} \rightarrow {}^{5}I_{8}$ (Ho³⁺), ${}^{3}H_{4} \rightarrow {}^{3}F_{4}$ (Tm³⁺), ${}^{3}F_{4} \rightarrow {}^{3}H_{6}$ (Tm³⁺) & ${}^{5}I_{7} \rightarrow {}^{5}I_{8}$ (Ho³⁺) transitions, respectively. The emission line width at 1 750 – 2 200 nm amounts to 250 nm, which is due to the overlap between ${}^{3}F_{4} \rightarrow {}^{3}H_{6}$ (Tm³⁺) and ${}^{5}I_{7} \rightarrow {}^{5}I_{8}$ (Ho³⁺). It is indicated that the crystal can be used as a high quality tunable laser medium at 2 μ m.



Fig. 2 Fluorescent spectrum at $900 - 2\ 300$ nm of the $KHo_{0.04}Tm_{0.06}Yb_{0.9}(WO_4)_2$ crystal

The energy level diagram sketched in Fig. 3 can describe the energy transfer among Tm^{3+} , Yb^{3+} , Ho^{3+} ions in the crystal. The fluorescence emission at 1 750 – 2 200 nm is due to the following three energy transmission interaction.

(1) Energy transfer of $Yb^{3+} \rightarrow Tm^{3+}$

Yb³⁺ ions can uptake quantities of pumping light to generate ${}^{2}F_{7/2} \rightarrow {}^{2}F_{5/2}$ absorption transition under 980 nm excitation. Then the energy was transferred from ${}^{2}F_{5/2}$ (Yb³⁺) to ${}^{3}H_{6}$ (Tm³⁺) and ${}^{3}H_{6} \rightarrow {}^{3}H_{5}$ transition of Tm³⁺ ions can be achieved. From ${}^{3}H_{5}$ energy state (Tm³⁺), the emission at 1 700 − 1 900 nm (${}^{3}F_{4} \rightarrow {}^{3}H_{6}$) can be obtained by the non-radiative transition process from ${}^{3}H_{5}$ to ${}^{3}F_{4}$ energy state.

(2) Energy transfer of $Yb^{3+} \rightarrow Ho^{3+}$

 ${\rm Yb}^{3\, *}$ ions generate a $^2F_{7/2} \rightarrow ^2F_{5/2}$ absorption transition due to the 980 nm excitation. Then the energy is transferred from $^2F_{5/2}(\,{\rm Yb}^{3\, *}\,)$ to $^5I_8(\,{\rm Ho}^{3\, *}\,)$ and $^5I_8 \rightarrow ^5I_6$ transition of ${\rm Ho}^{3\, *}$ ions can be achieved. From 5I_6 energy state (${\rm Ho}^{3\, *}\,)$, the emission at

1 900 – 2 200 nm(${}^{5}I_{7} \rightarrow {}^{5}I_{8}$) can be obtained by the non-radiative transition process from ${}^{5}I_{6}$ to ${}^{5}I_{7}$ energy state.

(3) Energy transfer of $\text{Tm}^{3+} \rightarrow \text{Ho}^{3+}$

This efficient resonant energy transfer happens due to the energy approximate overlap between the ${}^3F_4(\,\mathrm{Tm}^{3\,+}\,)$ and ${}^5I_7(\,\mathrm{Ho}^{3\,+}\,)$ energy levels. Therefore, the emission at 1 900 – 2 200 nm (${}^5I_7{\longrightarrow}{}^5I_8\,)$ can be achieved based on the energy transfer from ${}^3F_4(\,\mathrm{Tm}^{3\,+}\,)$ to ${}^5I_7(\,\mathrm{Ho}^{3\,+}\,).$



Fig. 3 Energy level diagram of Tm^{3+} , Yb^{3+} , Ho^{3+} ions and emission channels.

According to the formula $(4)^{[15]}$, the stimulated emission cross section of main emission peak at 2 030 nm can be calculated as 3.47×10^{-20} cm².

$$\sigma_{\rm em}(\lambda) = \frac{\lambda^4 I(\lambda)}{8\pi n^2 c \tau_{\rm f} \int I(\lambda) \, \mathrm{d}\lambda},\qquad(4)$$

where λ represents the fluorescent wavelength ($\lambda = 2\ 030\ \text{nm}$), *n* refers to the refractive index (*n* = 2.17), *c* represents the light speed in the vacuum ($c = 3 \times 10^8\ \text{m/s}$), $\tau_{\rm f}$ is the spontaneous fluorescence time ($\tau_{\rm f} = 1.8\ \text{ms}$). $I(\lambda)/\int I(\lambda) \,\mathrm{d}\lambda$ can be defined as the normalized line shape function, corresponding to ${}^{5}\mathrm{I}_{7} \rightarrow {}^{5}\mathrm{I}_{8}$ transition of Ho³⁺ ions.

4 Conclusion

KHo_{0.04} Tm_{0.06} Yb_{0.9} (WO₄)₂ laser crystal with a dimension of 26 mm × 11 mm × 9 mm was grown by the TSSG method. The testing result of absorption spectrum shows that Yb³⁺ ions have a wider absorption band at the range of 890 – 1 000 nm with a FWHM of 90 nm, and Tm³⁺ ions have a broader absorption band at 1 690 – 1 812 nm with a FWHM of 118 nm. The effective energy transfer of Yb→Ho, Yb→Tm, Tm→Ho can be easily achieved to produce 2 µm around laser due to these two broader

absorption bands at 890 – 1 000 nm (Yb³⁺) and 1 690 – 1 812 nm(Tm³⁺) in the crystal. The fluorescent spectrum measurement result indicates that the emission line width at 1 750 – 2 200 nm amounts to 250 nm, which is due to the overlap between ${}^{3}F_{4} \rightarrow {}^{3}H_{6}(Tm^{3+})$ and ${}^{5}I_{7} \rightarrow {}^{5}I_{8}(Ho^{3+})$. The spectral strength parameters of the crystal were calculated based on the Judd-Ofelt theory as follows: $\Omega_{2} =$ 10. 24 × 10⁻²⁰ cm², $\Omega_{4} = 2.37 \times 10^{-20}$ cm², $\Omega_{6} = 1.43 \times$ $10^{-20}\,{\rm cm}^2{}_{\circ}$ According to the energy level diagram of Tm³⁺, Ho³⁺ and Yb³⁺ ions, three kinds of energy transfer modes at 1 750 – 2 200 nm emission were discussed in the crystal. The stimulated emission cross section of main emission peak at 2 030 nm is 3.47 $\times 10^{-20}$ cm². It is indicated that KHo_{0.04}Tm_{0.06}-Yb_{0.9} (WO₄)₂ laser crystal can be widely used as an excellent laser gain medium at 2 μm around wavelength.

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毕琳(1984 -),女,吉林松原人,博 士,讲师,2016年于中国科学院大 学获得博士学位,主要从事计算机 应用、计算机模拟仿真等方面的 研究。

E-mail: bilin7080@163.com